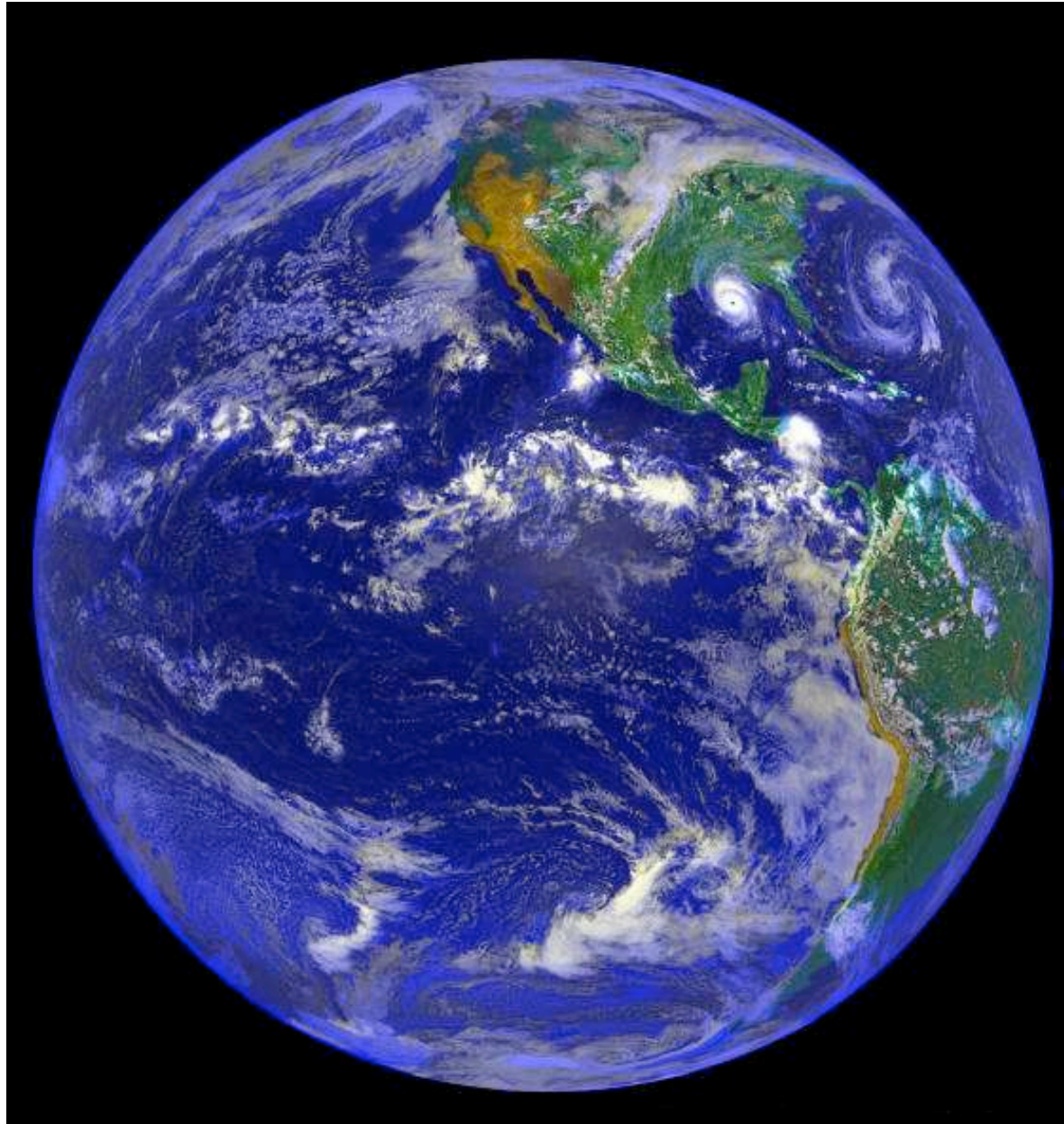


# OCEANOGRAPHY

## **1. Introduction to Planet “Earth”**

notes from the textbook, integrated with original contributions

Alessandro Grippo, Ph.D.



# Are there other Oceans in the Solar System?

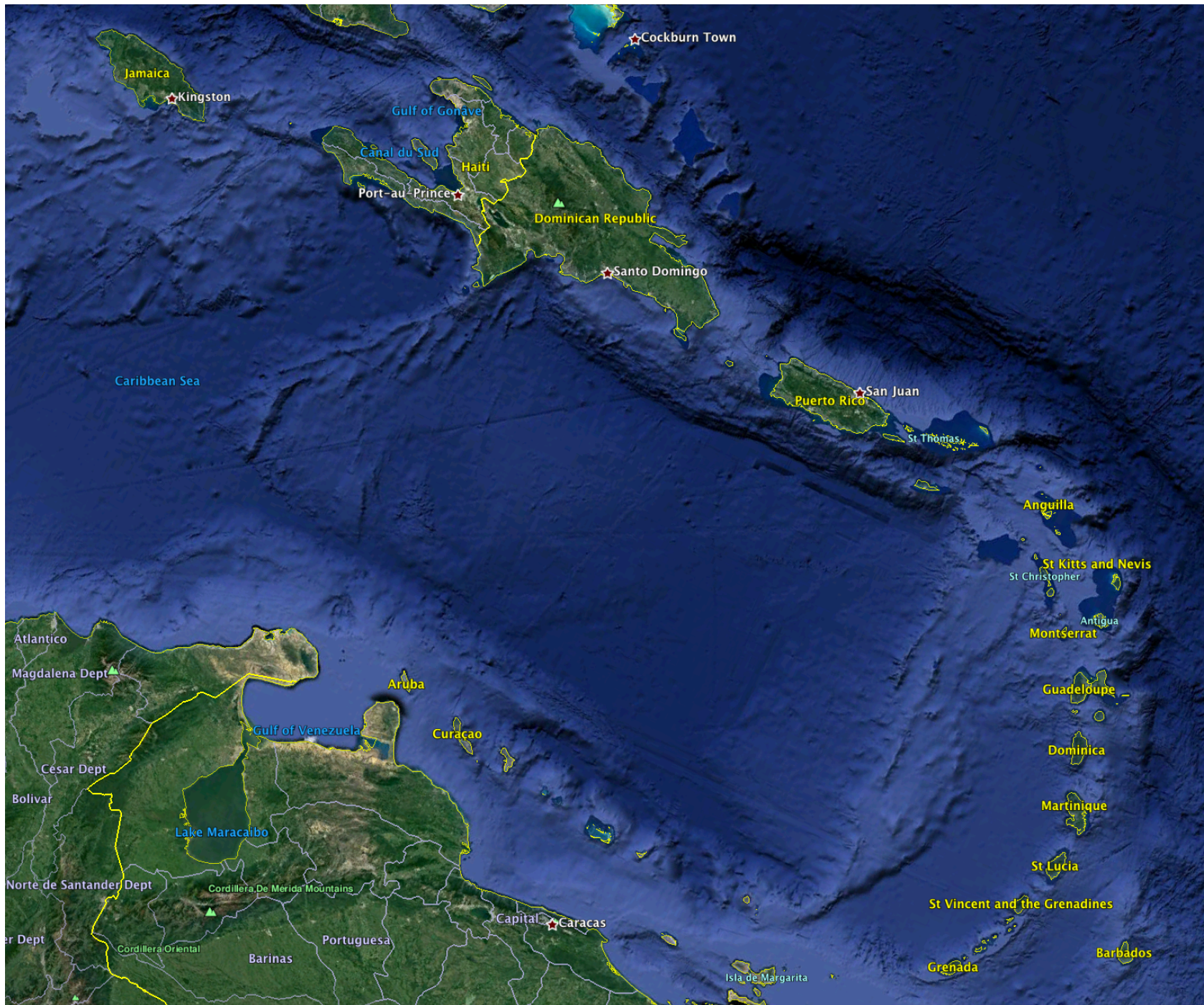
- Earth is the **only** planet where water is abundant, and it is in the liquid form
- Jupiter's moon
  - Europa has an icy surface and cracked structures
  - Ganimede, Callisto might have ice under their crust
- Saturn's moons
  - Enceladus has geysers of water vapor and ice
    - hydrothermal activity similar to that of Earth's oceans
  - Titan has oceans of liquid hydrocarbons at its surface

# Earth's Oceans

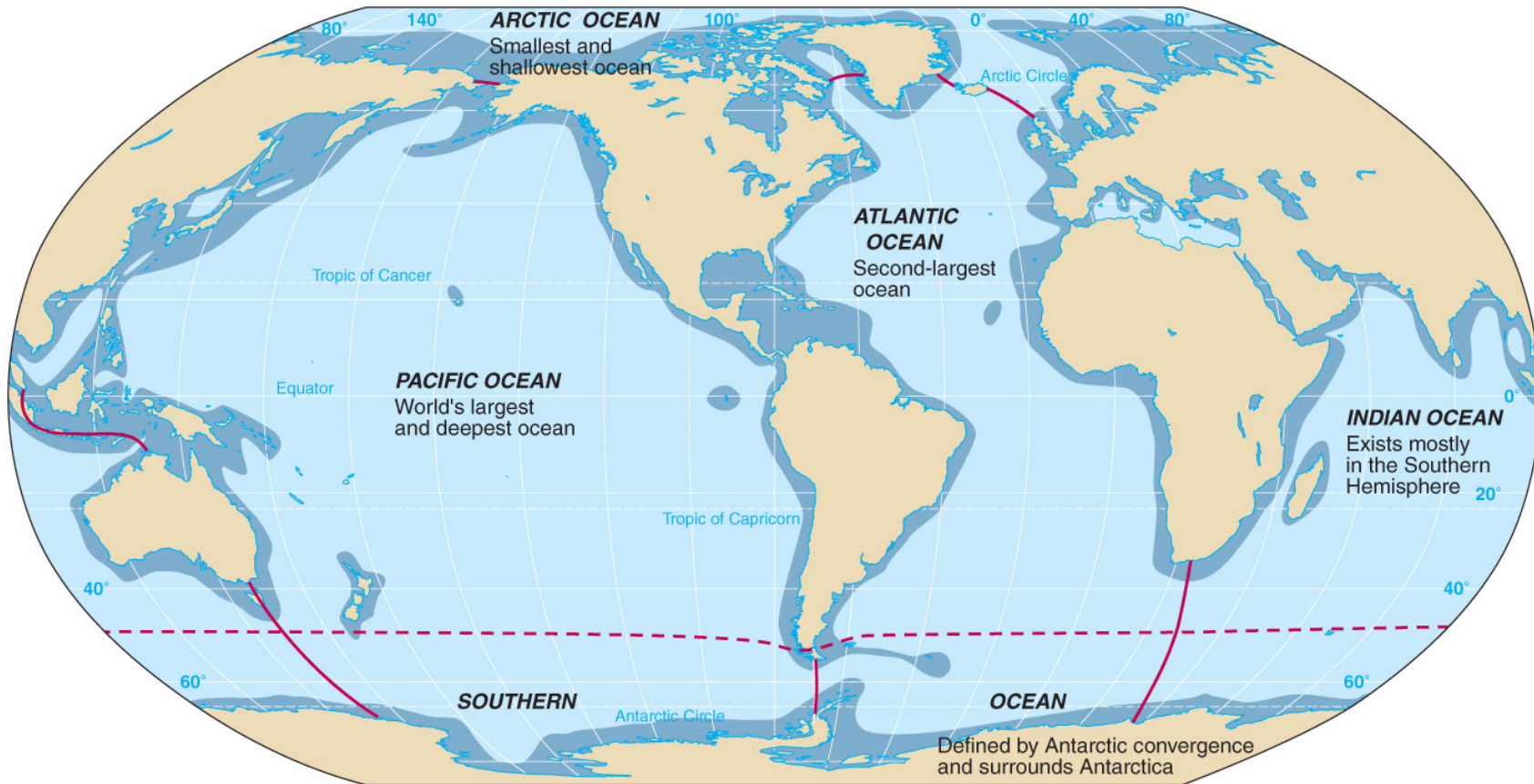
- essential for life today
- co-responsible for life development
- contain the greatest number of living organisms
- water is a major component of nearly every life-form
  
- influence climate and weather
  
- source of food, minerals, energy

# Earth's Oceans and us

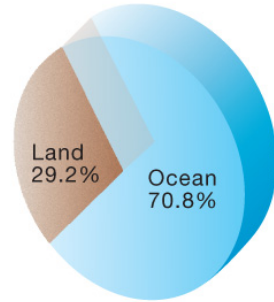
- oceans define geographical barriers
- oceans define political boundaries
- most people live in coastal areas
  - milder climate
  - cheaper transportation
  - pollution



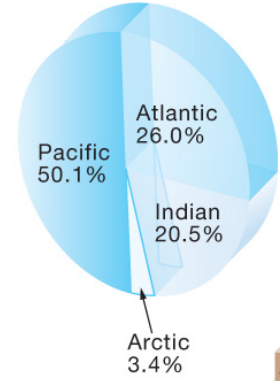
# Earth's Oceans



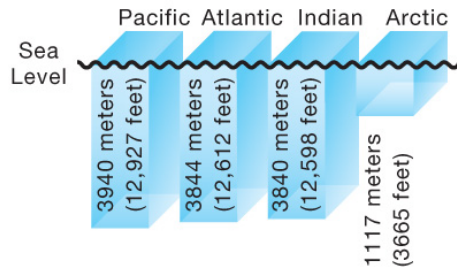
(a) Earth's Surface



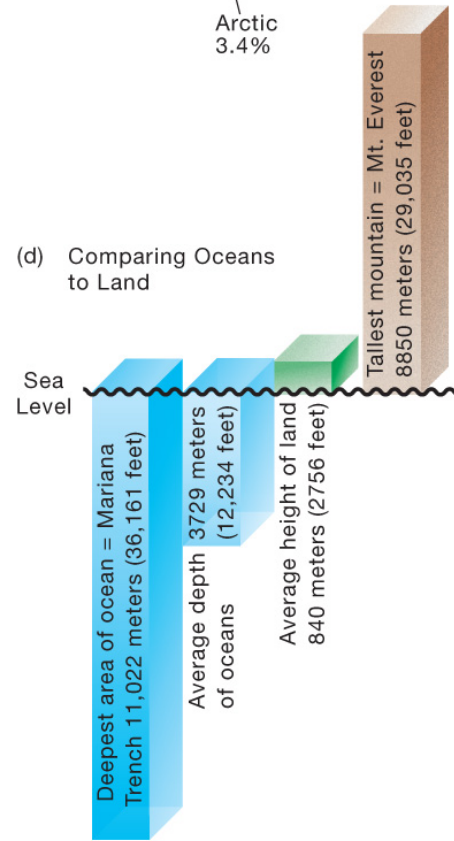
(b) Relative Ocean Size



(c) Average Ocean Depth

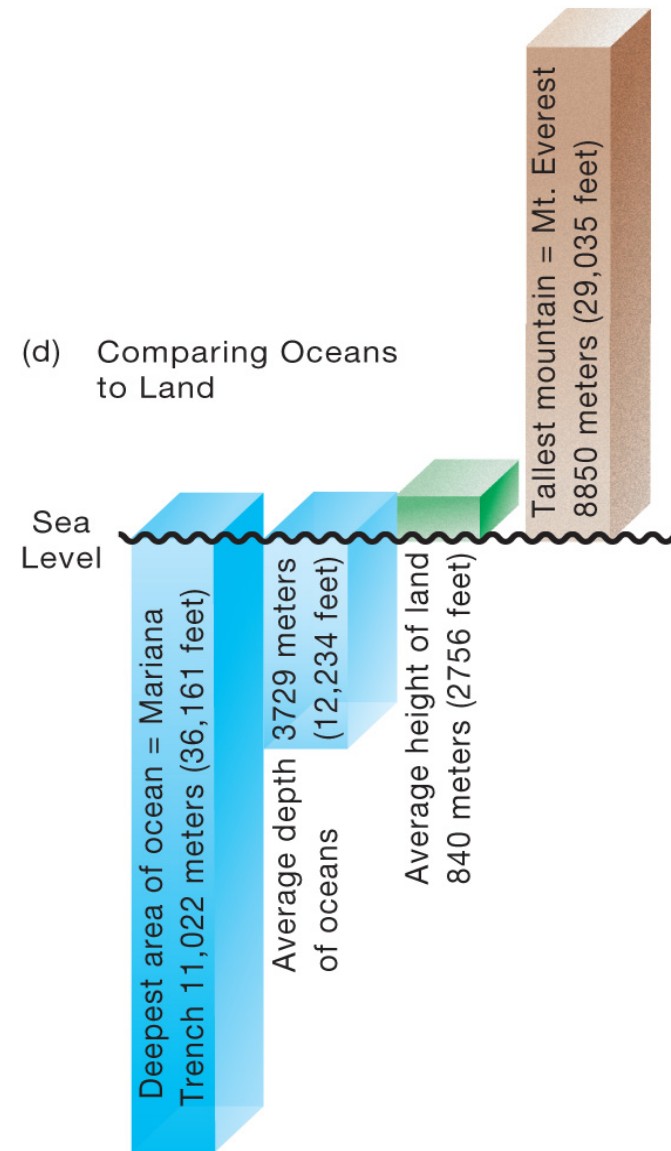


(d) Comparing Oceans to Land





- Average **ocean** depth is 3729 meters (12,234 feet)
- Average **continental** elevation is 840 meters (2756 feet)
- Deepest **ocean** trench is the Mariana Trench at 11,022 meters (36,161 feet)
- Highest **continental** mountain is Mt. Everest at 8850 meters (29,935 feet)

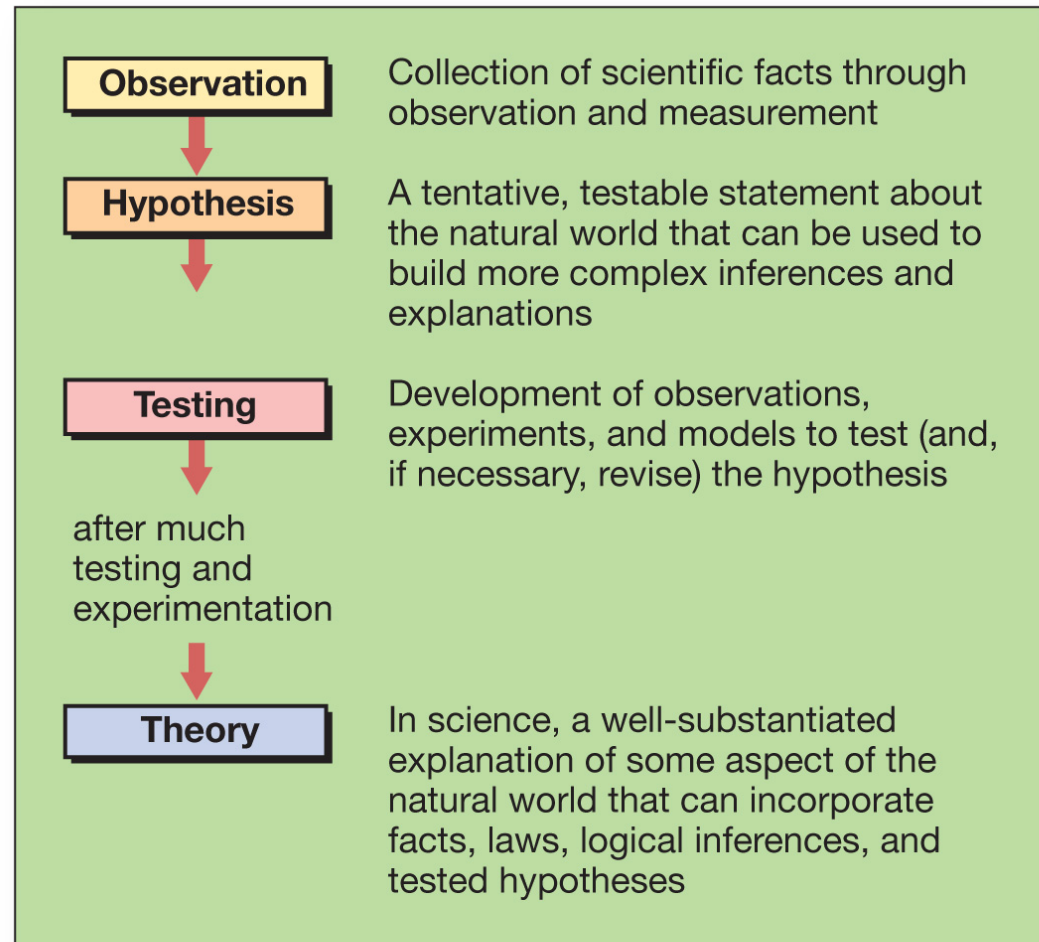


# Oceanography

- Scientific study of all aspects of the marine environment
- Four main disciplines:
  - Geological Oceanography
  - Chemical Oceanography
  - Physical Oceanography
  - Biological Oceanography

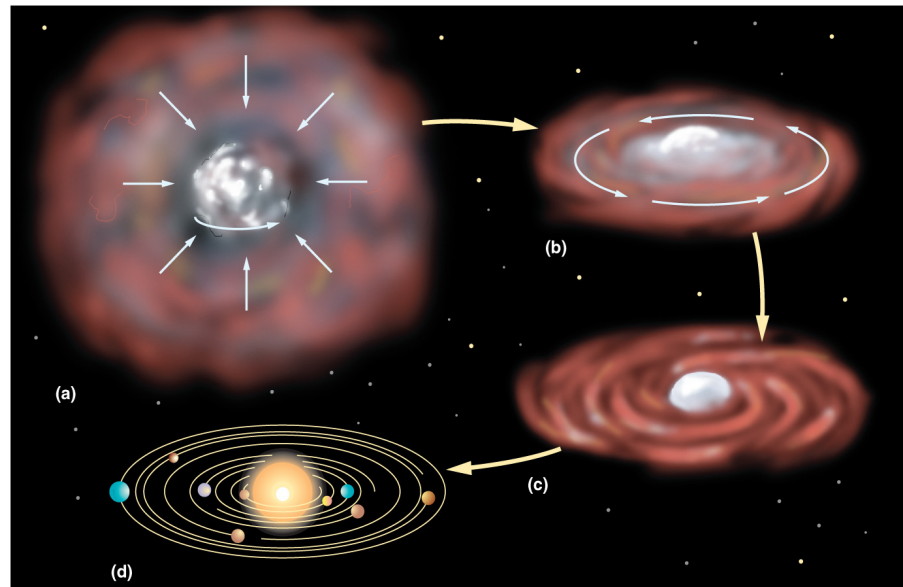
# Nature of Scientific Inquiry

- Natural phenomena governed by physical processes
- Physical processes similar today as in the past
- Scientists discover these processes and make predictions
- Called the **scientific method**



# Formation of Earth and the Solar System

- **Nebular hypothesis** – all bodies in the solar system formed from nebula
  - **Nebula** = cloud of gases and space dust
    - Mainly hydrogen and helium
- Gravity concentrates material at center of cloud (Sun)
- Protoplanets form from smaller concentrations of matter (eddies)



# Protoearth

- Larger than Earth today
- Homogeneous composition
- Bombarded by meteorites
  - Moon formed from collision with large asteroid
- Radioactive heat
  - Spontaneous disintegration of atoms
  - Fusion reactions
- Heat from contraction (protoplanet shrinks due to gravity)
- Protoearth partially melts
- Density stratification (layered Earth)

# EARTH'S SYSTEM TODAY

- **A rocky physical body**
  - Core, Mantle, Oceanic Crust, Continental Crust
- **The Water**
  - Hydrosphere
    - Salt waters: Oceans
    - Fresh waters: Glaciers, Groundwater, Rivers, Lakes
    - Water vapor in the atmosphere
- **The Air**
  - Atmosphere
    - 79% N<sub>2</sub>, 21% O<sub>2</sub>, traces of CH<sub>4</sub>, CO<sub>2</sub>, NO<sub>2</sub>, SO<sub>2</sub>, H<sub>2</sub>O and other gases
- **Living organisms**
  - Prokaryotes (Archaea, Bacteria)
  - Eukaryotes (Protists, Fungi, Plants, Animals)

## Where do the components come from?

- Solid **ROCKS** formed when Earth cooled
- Liquids and gases were originally released during volcanic eruptions (**WATER** and **ATMOSPHERIC GASES**)
  - Water and air were retained thanks to gravitational force
- **LIFE** started as soon as Earth's surface was solid

# Density Stratification

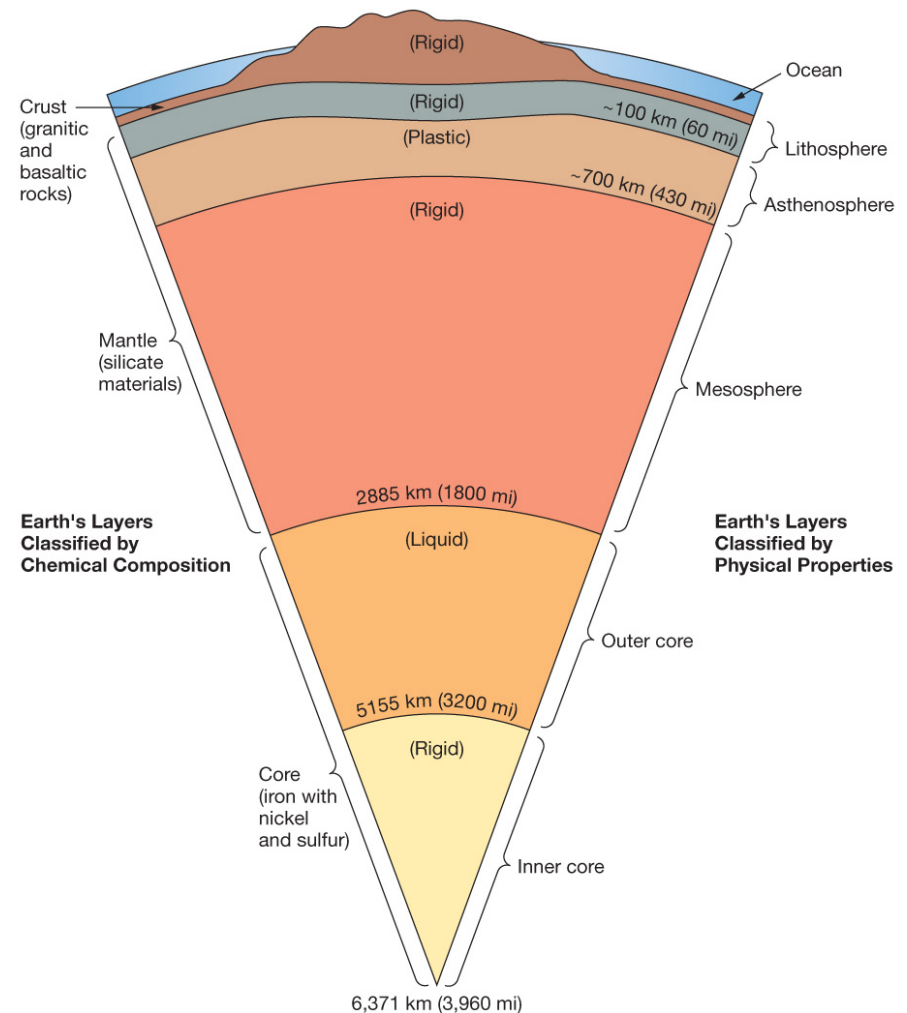
- High density = heavy for its size
- Early Earth experienced gravitational separation.
  - High density materials (Iron and Nickel) settled in core.
  - Less dense materials formed concentric spheres around core.



# Earth's Internal Structure

Layers defined by

- Chemical composition
- Physical properties



# Layers by Chemical Composition

- **Crust**
  - Low-density, mainly silicate minerals
- **Mantle**
  - Mainly iron (Fe) and magnesium (Mg) silicate minerals
- **Core**
  - High-density, mainly iron (Fe) and nickel (Ni)

# Layers by Physical Properties

- Lithosphere
  - Cool, rigid shell
  - Includes crust and upper mantle
  - About 100 km (60 miles) thick
- Asthenosphere
  - Relatively hot, plastic
  - Flows with high viscosity
  - Base of lithosphere to about 700 km (430 miles) deep
- Mesosphere
- Outer core
- Inner core

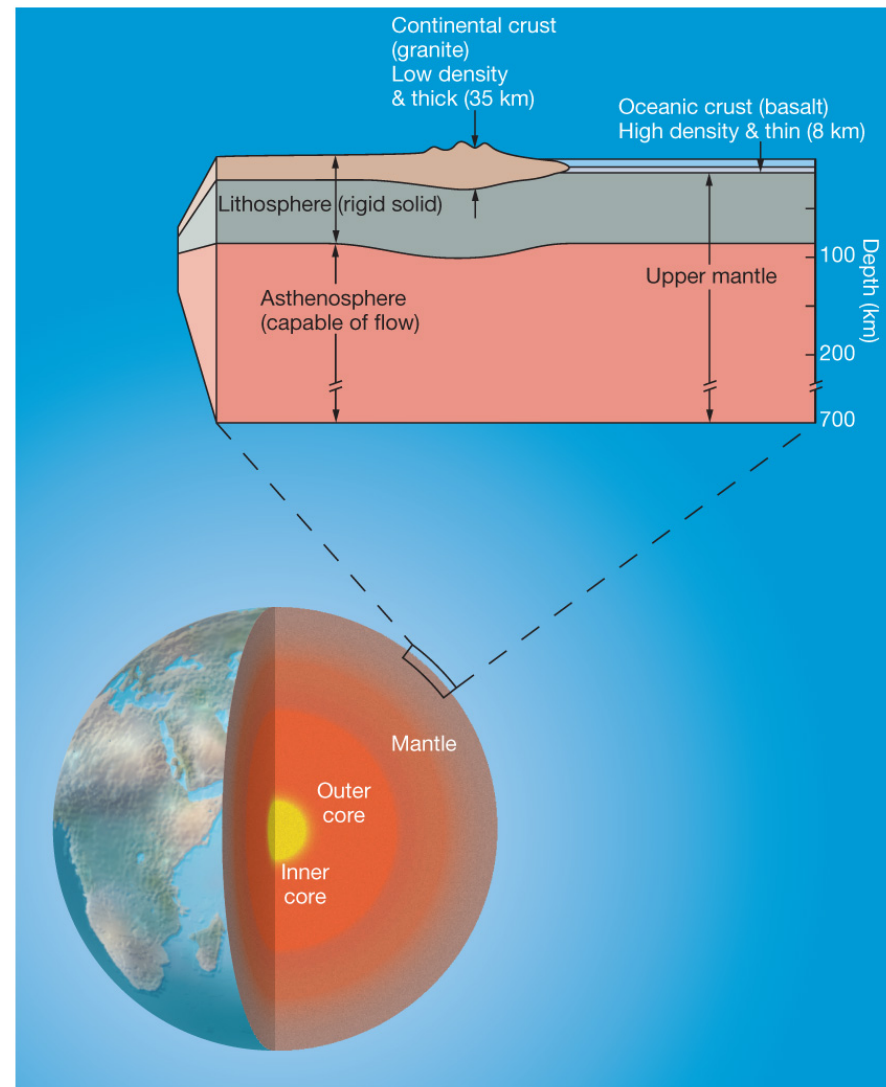


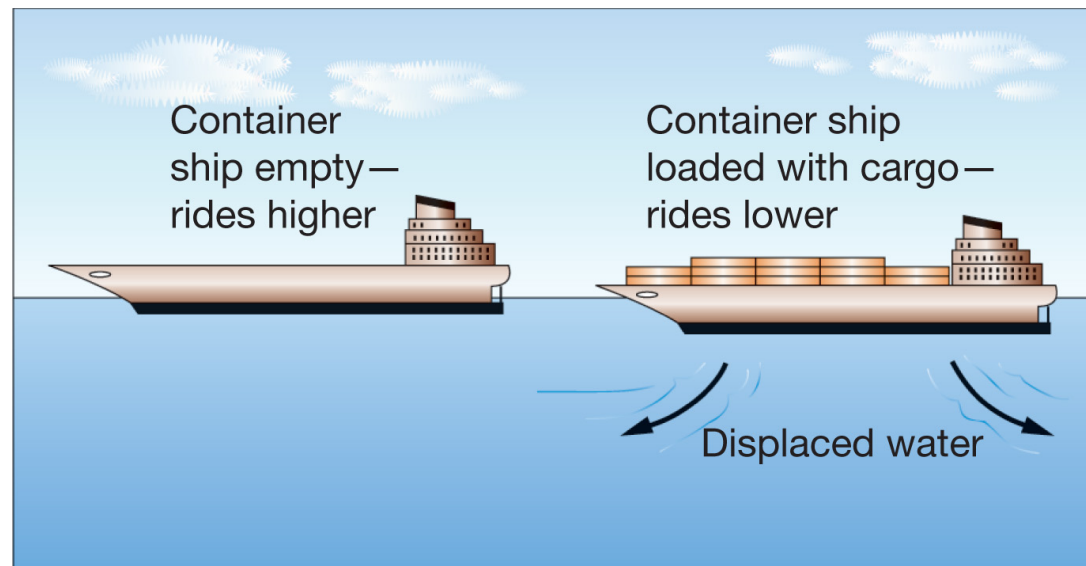
TABLE 1.1

## COMPARING OCEANIC AND CONTINENTAL CRUST

	<b>Oceanic crust</b>	<b>Continental crust</b>
Main rock type	Basalt (dark-colored igneous rock)	Granite (light-colored igneous rock)
Density (grams per cubic centimeter)	3.0	2.7
Average thickness	8 kilometers (5 miles)	35 kilometers (22 miles)

# Isostatic Adjustment

- Vertical movement of Earth's crust
- Buoyancy of lithosphere on asthenosphere
  - Less dense continental crust floats higher than denser oceanic crust
- **Isostatic rebound** – rising of crust formerly weighed down by glacier ice



# WATER

- In Oceans
  - salt water; salinity is variable but the types and ratios of salt are identical all over the world in open oceans; mostly  $\text{Na}^+$  and  $\text{Cl}^-$  ions



The Pacific Ocean: Howe Sound, by Vancouver, B.C. Canada

# WATER

- In Rivers
  - fresh water; composition may vary, depending on source



**The Virgin River in Zion National Park  
Springdale, Utah**

# WATER

- In Lakes
  - fresh water or salt water ; if salty, type of salt may vary



**An ephemeral, saline lake: Badwaters  
Death Valley National Park, California**



# WATER



**Convict Lake, Sierra Nevada, California**

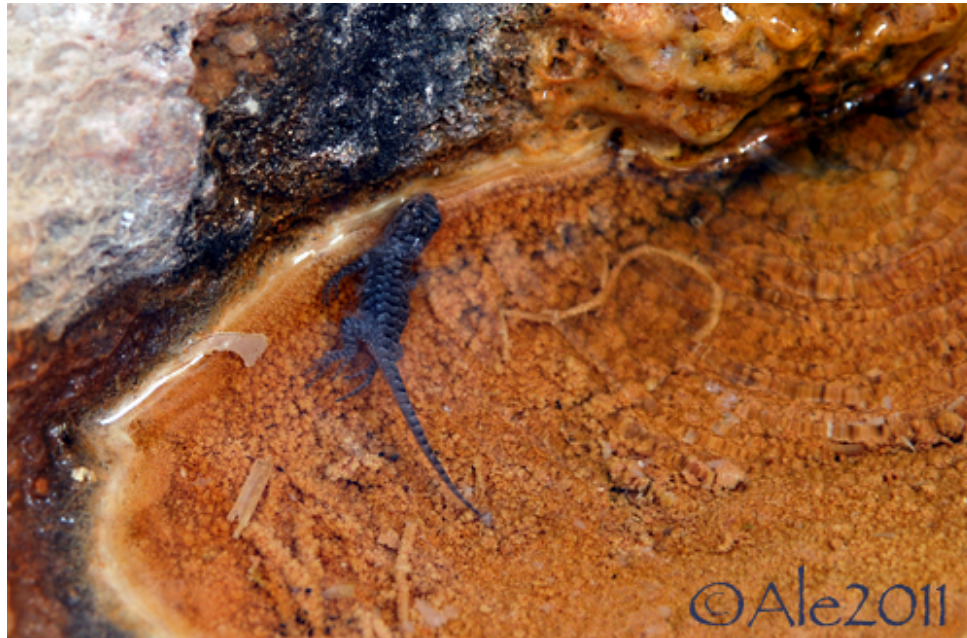
# WATER



**Mono Lake, Lee Vining, California**

# WATER

- As Groundwater
  - fresh water; can be salty close to ocean or at depth
  - temperature may vary: geothermal waters



Hot groundwater upwells at this location, killing animals who fall into pools, and causing deposition of calcium carbonate (orange crystals)

**Hot Springs by Bridgewater, California**

# WATER



Hot water pools in **Yellowstone National Park, WY**  
the colors are the result of different colonies of bacteria that thrive in  
progressively lower water temperatures

# WATER



A geyser in Yellowstone National Park, WY

# WATER

- As Ice
  - Ice on continents
  - Ice pack (sea ice), can contain up to 0.8% salt
  - Alpine ice



**Glacier Bay, Gustavus, Alaska**

# WATER



clockwise from upper left:

the Andes at the Chile/Argentina border; the Alaska Range in Seward, AK;  
a glacial cirque in the Rocky Mountains, CO; a glacier in the Coast Mountains, AK

# WATER

- As Water Vapor
  - Atmospheric moisture, usually “distilled” water
  - Percentage of water in atmosphere can vary



Clouds over the island of **Maui, Hawaii**

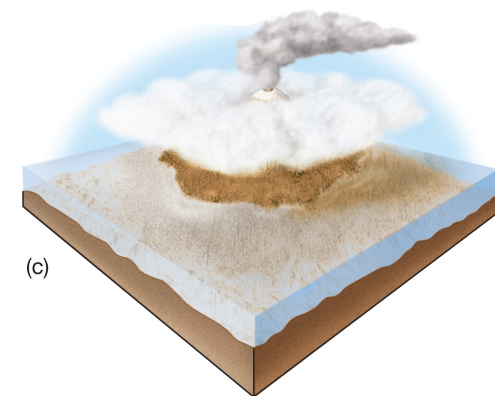
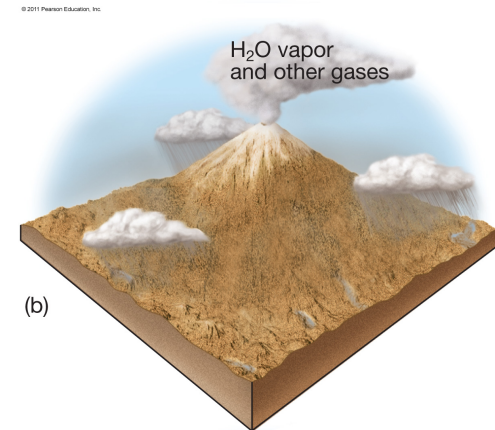
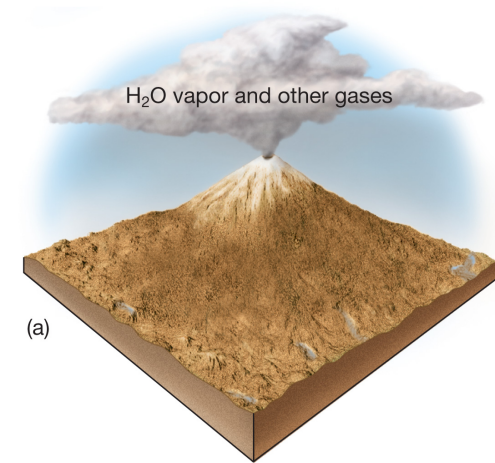


# WATER

- In Rocks
  - directly as water (e.g. Gypsum,  $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ )
  - trapped in clay minerals
  - as  $\text{OH}^-$  ions that can promptly react with  $\text{H}^+$  ions
- In Living Organisms
  - Plants (evaporation and transpiration processes)
  - Animals

# Origin of Earth's Oceans

- Outgassed water vapor fell as rain.
- The first permanent oceans formed 4 billion years ago.
- Salinity developed from dissolved rock elements.
  - Early acidic rain dissolved more crustal minerals than today.



# Origin of Earth's Atmosphere

- **Outgassing** – occurred during density stratification
  - Water vapor
  - Carbon dioxide
  - Hydrogen
  - Other gases
- Earth's early atmosphere different from today

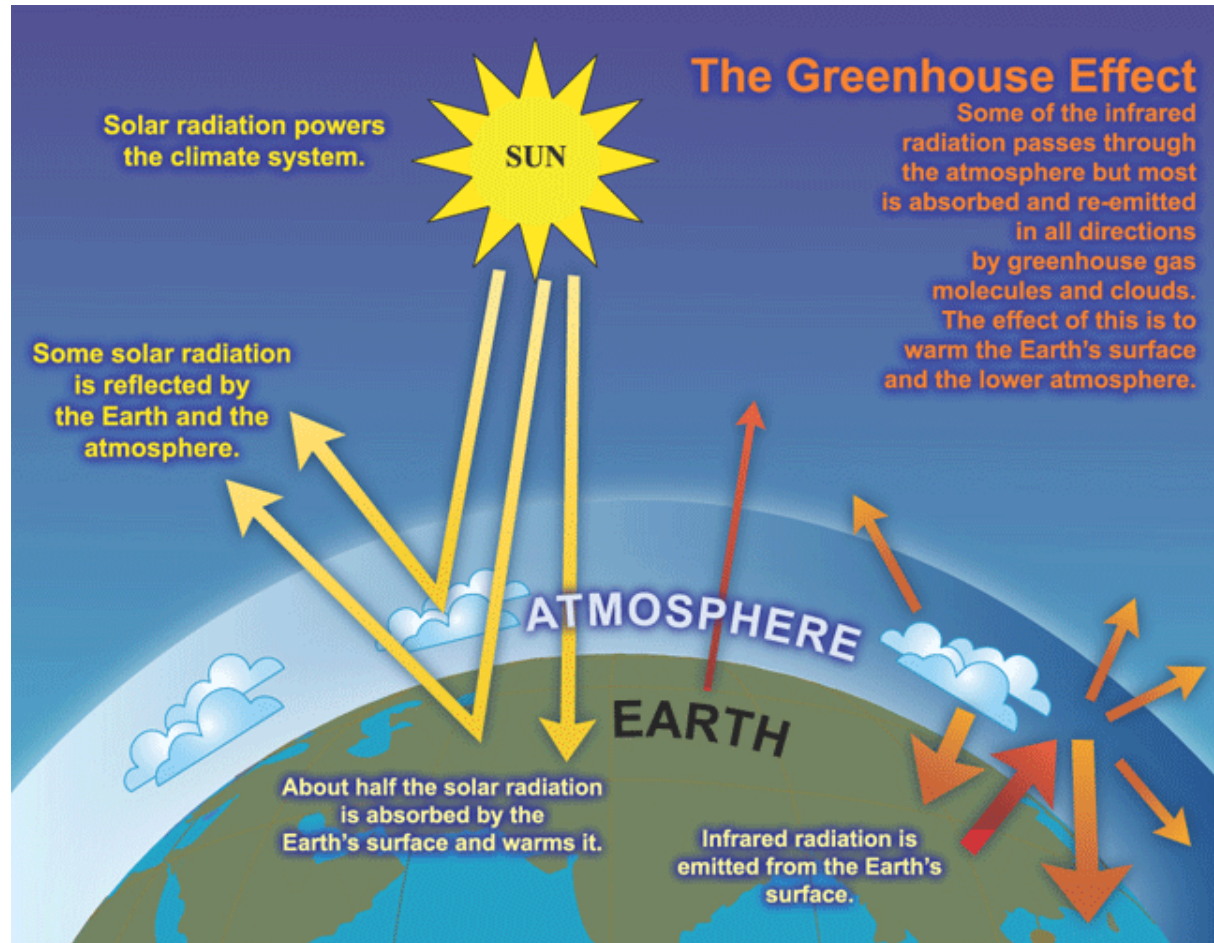
# AIR

- Nitrogen: 78%
- Oxygen: 21%
- other gases: ~ 1%
  - ( $\text{CO}_2$ ,  $\text{NO}_2$ ,  $\text{CO}$ ,  $\text{NO}$ ,  $\text{CH}_4$ ,  $\text{H}_2\text{O}$ , and others)

in orange: greenhouse gases

# AIR

- Greenhouse gases:
  - Solar radiation can penetrate greenhouse gases
  - Radiation is reflected back by Earth, but at a different frequency
  - Greenhouse gases trap the radiation emitted by Earth
  - More greenhouse gases in the atmosphere imply warmer temperatures



from [www.co2now.org](http://www.co2now.org)

# Life's Possible Ocean Origins

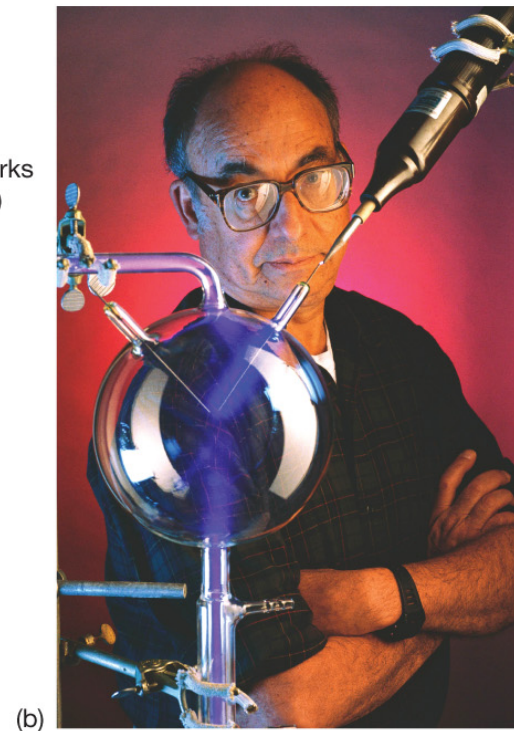
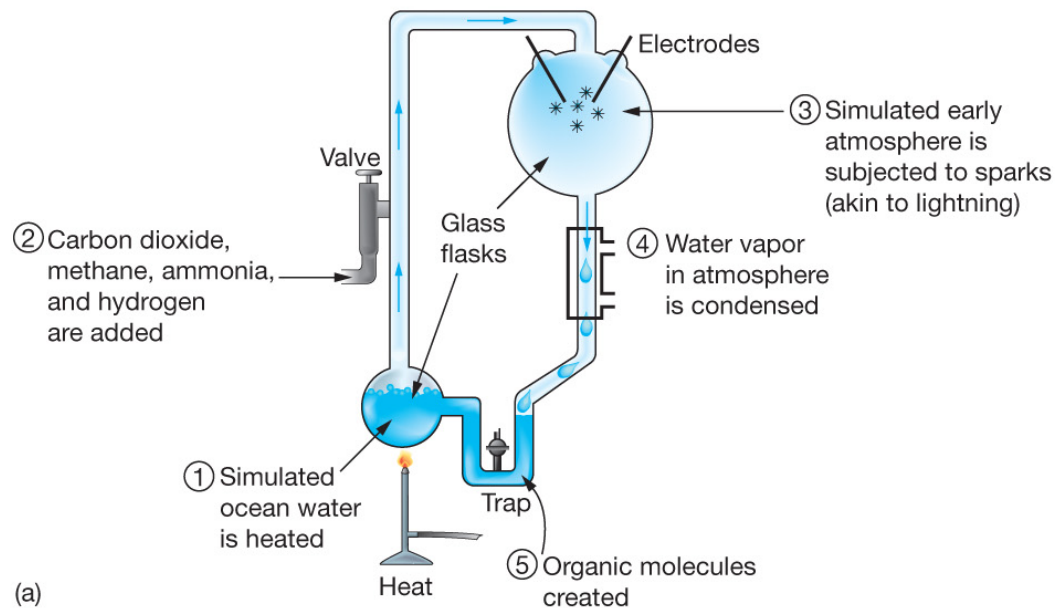
- Earth's earliest known life forms are 3.5-billion-year-old bacteria fossilized in ocean rocks.
- Started as:
  - chemosynthetic
  - prokaryotic
  - mitotic
  - single celled

- Evolved in time into:
  - photosynthetic
  - eukaryotic
  - meiotic (sexual reproduction)
  - colonial
  - from colonial to multicellular
  - from simple (e.g. sponges) to specialized (e.g. plants and animals)



# Stanley Miller's Experiment

- Organic molecules formed by ultraviolet light, electrical spark (lightning), and a mixture of water, carbon dioxide, hydrogen, methane, and ammonia



- **Heterotrophs**

- Very earliest life
- Require external food supply

- **Autotrophs**

- Evolved later
- Manufacture own food supply

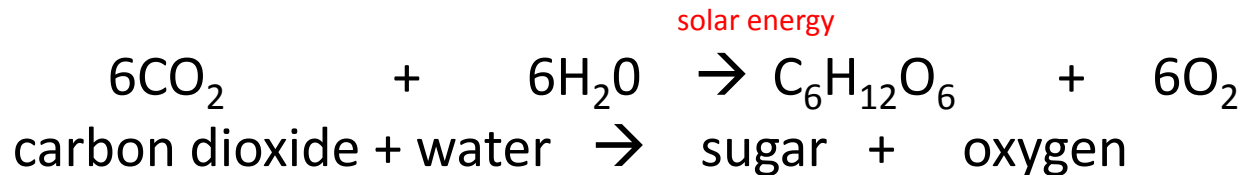
# First Autotrophs

- Probably similar to modern **anaerobic** bacteria
  - Survive without oxygen
- **Chemosynthesis** from chemicals at deep hydrothermal vents
- Supports idea of life's origins on deep ocean floor in absence of light

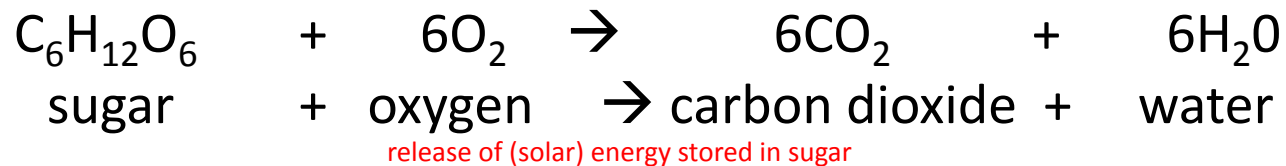
# Phptosynthesis and Respiration

- Complex autotrophs developed **chlorophyll**.
- This allowed the use of the Sun for **photosynthesis**.
- **Cellular respiration**

- Photosynthesis



- Respiration



- All organisms modify the environment
  - Photosynthesis
    - added oxygen to atmosphere
    - oxygen provides more energy as a fuel
    - life development sped up
  - Carbonate reefs
    - today they are limited, more developed in the past
  - Human activity

# Oxygen

- Life requires  $O_2$ .
- Ozone ( $O_3$ ) protects from ultraviolet radiation.
- Early Earth had little free oxygen.
- The lack of ozone may have helped originate life.

- Destruction vs. Preservation of Organic Matter
  - oxygen is needed to burn sugars
  - upon death, bodies are recycled by scavengers, decomposers, who all need oxygen
  - if there is no oxygen, organic matter (the sugars) do not decompose and are preserved, albeit modified
  - this preserved organic matter that escaped decomposition makes for hydrocarbons (oil and natural gas)

# Great Oxidation Event

- 2.45 billion years ago
- Increased oxygen and ozone eliminated the anaerobe food supply.
- Light and oxygen kill anaerobes.
- Cyanobacteria adapted and thrived.

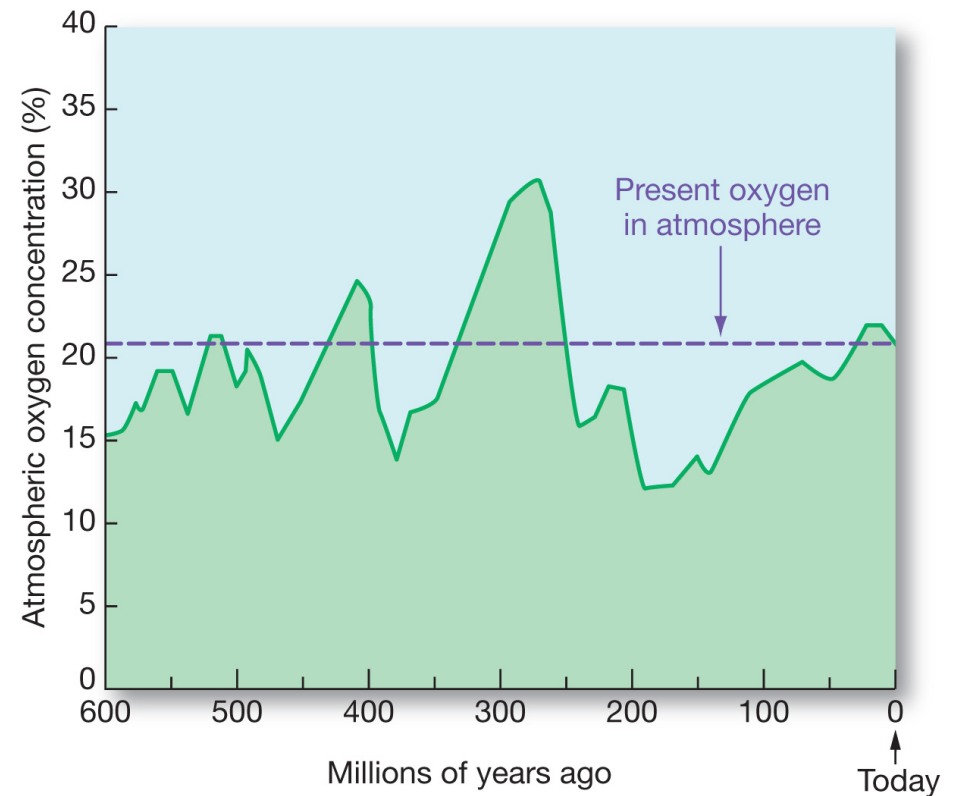


# Evolution and Natural Selection

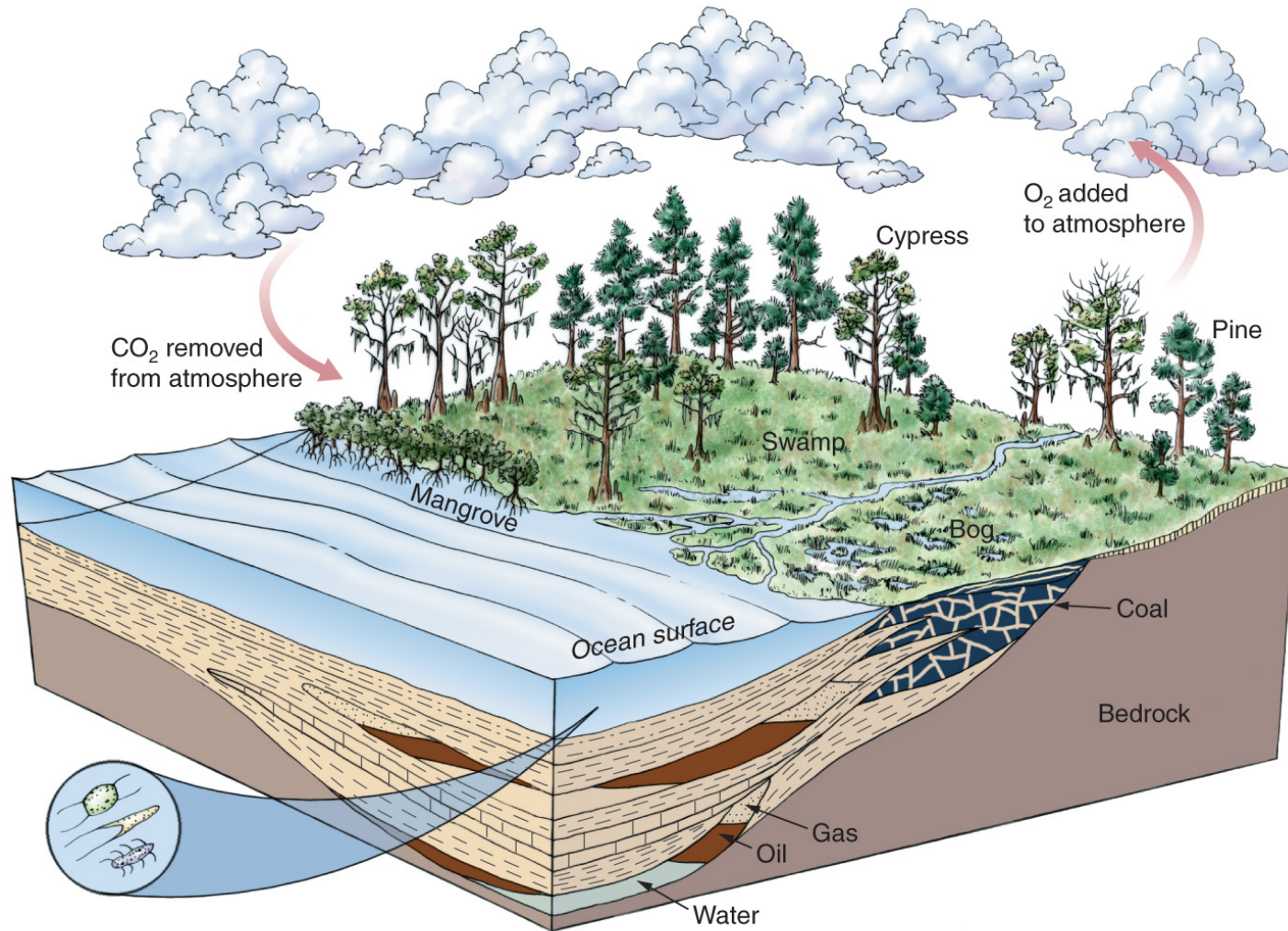
- Organisms adapt and change through time.
- Advantageous traits are naturally selected.
- Traits are passed to the next generation.
- Organisms adapt to environments.
- Organisms can modify environments.

# Changes to Earth's Atmosphere

- Photosynthetic organisms are responsible for life as we know it today.
- Reduce CO<sub>2</sub>, increase O<sub>2</sub> to 21%
- High oxygen = biodiversity increase
- Low oxygen associated with extinction events



# Algae, Plants, and Earth's Environment



# Age of Earth

## Deep time

- Geology shows that Earth is 4.6 billion years old
  - Humans have been around for an incredibly smaller amount of time
- 
- Relative Time
  - Numerical Time

# DEEP TIME

- Relative Time approach
  - tells us the order of event, not their age
  - basic principles make it easy
  - can be done visually in the field
- Numerical Time approach
  - tells us the numerical age of rocks and/or events
  - can only be done in the lab
  - it only works on certain rocks
  - it is expensive and time-consuming

